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Writing the Net Ionic Reaction $\text{Na}_3\text{PO}_4 + \text{Sb}(\text{NO}_3)_5 \rightarrow \text{NaNO}_3 + \text{Sb}_3(\text{PO}_4)_5$ Example Reaction 1) Step one BALANCE the overall chemical reaction. $5 \text{Na}_3\text{PO}_4 + 3 \text{Sb}(\text{NO}_3)_5 \rightarrow 15 \text{NaNO}_3 + \text{Sb}_3(\text{PO}_4)_5$ 2) Use the PPT Table or physically what is happening to determine the phases. $5 \text{Na}_3\text{PO}_4(\text{aq}) + 3 \text{Sb}(\text{NO}_3)_5(\text{aq}) \rightarrow 15 \text{NaNO}_3(\text{aq}) + \text{Sb}_3(\text{PO}_4)_5(\text{s})$ 3) Bring down whatever compound is a solid, liquid, gas, or weak electrolyte in ...

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